Oxford A Level Sciences
OCR Physics B

| Question | Answer | Marks |
| :---: | :---: | :---: |
| Section A |  |  |
| 1 a | (A gas) in which some atoms/molecules/particles have lost electrons to become positive ions/charged ions. | $\begin{array}{\|l} \hline 1 \\ 1 \\ \hline \end{array}$ |
| 1 b | $\lambda=\frac{c}{f}=1.0(3) \times 10^{-7} \mathrm{~m}$ | 1 |
| 1 c | $\begin{aligned} & E=V Q=240 \times 1.6 \times 10^{-19} \\ & =3.8(4) \times 10^{-17} \mathrm{~J} \end{aligned}$ | $\begin{array}{\|l\|} \hline 1 \\ 1 \\ \hline \end{array}$ |
| 1di1 | 240 V | 1 |
| 1di2 | 120 V | 1 |
| 1 dii | $\begin{aligned} & \text { From the graph, 0.26(4) } \mu \mathrm{A} \\ & \text { Power }=2.64 \times 10^{-7} \times 180 \times 6.2 \times 10^{6} \\ & =295 \mathrm{~W} \end{aligned}$ | $\begin{array}{\|l\|} \hline 1 \\ 1 \\ 1 \end{array}$ |
| 2 ai | There are many small crystals with close-packed planes of different alignments/with grain boundaries. | $\begin{array}{\|l\|} \hline 1 \\ 1 \\ \hline \end{array}$ |
| 2 aii | Ductile means it can be drawn into a wire. <br> Two marks for any two further points, e.g.: <br> - In the pure metal, there is a dislocation in the regular crystal where atoms are free to move. <br> - Slips occur easily where planes move over each other. | $\begin{array}{\|l\|} \hline 1 \\ 2 \\ \hline \end{array}$ |
| 2 bi 1 | Hard materials are difficult to dent or scratch (they resist wear). | 1 |
| 2bi2 | Hard materials last longer / don't blunt so easily / give a cleaner or more accurate cut. | 1 |
| 2 bii | Any two points from e.g.: <br> - Metals have free/delocalised electrons. <br> - Metals have non-directional bonds <br> - which hold positive ions in lattice / allow positive ions to slip or dislocate. <br> Any two points from e.g.: <br> - In diamond, there are bound/localised electrons <br> - which form strong directional bonds. <br> - Diamond forms a giant lattice that is hard to displace. | 2 2 |
| 3ail | $2.0 \mathrm{~km} \mathrm{~h}^{-1}$ | 1 |
| 3ai2 | $8.0 \mathrm{~km} \mathrm{~h}^{-1}$ | 1 |
| 3 aii | $\begin{aligned} & \text { Time }=\frac{\text { distance }}{\text { velocity }} \\ & \text { Time from } \mathbf{A} \text { to } \mathbf{B}=\frac{2 \mathrm{~km}}{(5-3) \mathrm{km} \mathrm{~h}^{-1}}=1 \mathrm{~h} \\ & \text { Time from } \mathbf{B} \text { to } \mathbf{A}=\frac{2 \mathrm{~km}}{(5+3) \mathrm{km} \mathrm{~h}^{-1}}=0.25 \mathrm{~h} \\ & \text { Total time }=1+0.25=1.25 \mathrm{~h} \mathrm{(1.3} \mathrm{~h} \mathrm{to} 2 \text { s.f. }) \end{aligned}$ | 1 (for method) <br> 1 |
| 3 bi | Either correct scale drawing OR $\begin{aligned} & \sin ^{-1}\left(\frac{3.0}{5.0}\right) \\ & =36.8 \ldots=37^{\circ}(2 \text { s.f. }) \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 2 \\ & 1 \\ & 1 \\ & 1 \end{aligned}$ |
| 3 b ii | $\begin{aligned} & \text { resultant velocity }=\sqrt{3.0^{3}+5.0^{2}} \\ & =4.0 \mathrm{~km} \mathrm{~h} \\ & \text { (alternatively using sin or } \cos 37^{\circ} \text { and a known velocity) } \end{aligned}$ | $\begin{array}{\|l\|} \hline 1 \\ 1 \\ \hline \end{array}$ |

Oxford A Level Sciences
OCR Physics B

| 3 c | Time taken for $\mathbf{Q}$ to reach $\mathbf{C}=\frac{\text { distance }}{\text { velocity }}=\frac{2.0}{4.0}=0.50 \mathrm{~h}$ Therefore distance travelled by $\mathbf{P}=0.5 \times 2.0=1.0 \mathrm{~km}$ $\theta=\tan ^{-1} \frac{1.0}{2.0}=26.5 \ldots=27^{\circ}(2$ s.f. $)$ | 1 1 |
| :---: | :---: | :---: |
| 4 ai | Straight line passing through the origin of half the gradient of A. | 1 |
| 4 aii | Straight line passing through of 4 times the gradient of A . | 1 |
| 4 b i | (Direct) proportionality/straight line through origin. | 1 |
| 4 b ii | $\begin{aligned} & \sigma=\frac{1.2(8) \times 10^{8}}{1.3 \times 10^{8}} \\ & \varepsilon=0.00070 \\ & E=1.8(3) \times 10^{-11} \mathrm{~Pa} \end{aligned}$ | 1 1 1 |
| 4 c | Any 4 sensible points, e.g.: <br> - Metal atoms lose electrons/form positive ions. <br> - Free/delocalised (negative) electrons are present. <br> - Attractive forces/non-directional bonding between positive and negative ions cause bonds. <br> - Positive metal ions are closely-packed/regularly stacked in planes or lattices. <br> - When positive ions are given small displacement, atomic planes move relative to neighbours. <br> - Ions return to their positions when the displacing force is removed. | 4 |
| Section B |  |  |
| 1 a | $\text { power }=\frac{1}{4.5 \times 10^{-3}}=222 \mathrm{D}$ | 1 |
| 1 a ii | $\begin{aligned} & \frac{1}{v}=-2+222=220 \mathrm{D} \\ & v=4.5(4) \times 10^{-3} \mathrm{~m} \end{aligned}$ | 1 1 |
| 1 a iii | $\begin{aligned} & \frac{v}{u}=\frac{4.54 \times 10^{-8}}{0.5} \\ & =9.1 \times 10^{-3} \end{aligned}$ | 1 1 |
| 1 b | Multiply both dimensions by $9.1 \times 10^{-3}$ gives dimensions $2.4 \times 2.0 \mathrm{~mm}$ |  |
| 1 c | $\begin{aligned} & \text { Resolution }=\frac{270}{1200} \\ & =0.23 \mathrm{~mm} \end{aligned}$ <br> This is bigger than the width of an eyelash so if eyelashes are not separated by at least a gap of about one eyelash thick, they will not be resolved. | 1 1 1 |
| 1 di | 256 levels are coded by 8 bits, which gives $3 \times 8=24$ bits per pixel bits in one image $=24 \times 1200 \times 1000=2.88 \times 10^{7}$ <br> number of images $=\frac{0.9 \times 10^{9}}{2.88 \times 10^{7}}=31$ images | 1 1 1 |
| 1 dii | decrease bits per image by e.g. fewer intensity levels but this will decrease image quality/accuracy of the match with object. | 1 1 1 |

Oxford A Level Sciences
OCR Physics B

| 2 a | Marks are awarded for a well-structured answer. Relevant points can include e.g.: <br> - At low temperatures, NTC thermistors have a very large resistance whilst PTC thermistors have a reasonably low resistance. <br> - At high temperatures, NTC thermistors have low (next to no) resistance whilst PTC thermistors have a very large resistance. <br> - NTC thermistors display a steady exponential decline in resistance with increasing temperature. <br> - PTC thermistors have a relatively steady resistance, with large and sudden increase at a given temperature. <br> - Reasons: e.g. The molecules/atoms in an NTC thermistor require energy to allow electrons to pass through the material. <br> - Reasons: e.g. The molecules/atoms in a PTC resistor change structure at a given temperature/energy so that no electrons can pass freely through to material. | 6 |
| :---: | :---: | :---: |
| 2 b | The PTC thermistor shows a large range of resistance over a small temperature range (over $100^{\circ} \mathrm{C}$ ), but little change around $50^{\circ} \mathrm{C}$, so it would not be suitable for keeping something at $50^{\circ} \mathrm{C}$. <br> The NTC thermistor shows a steady decreases in resistance over this temperature range, so it would be suited to the task | 1 |
| 2 ci | $\begin{aligned} & P=I V=\frac{V^{2}}{R} \\ & =\frac{6.0^{2}}{100} \\ & =0.36 \mathrm{~W} \end{aligned}$ | 1 1 1 |
| 2 cii | When the current is switched on, the temperature rises. This rise causes a slight decrease in resistance. At around $50^{\circ} \mathrm{C}$ the resistance remains steady. So the current will remain steady. | 1 1 1 1 |
| Section C |  |  |
| 1 ai | Measuring from the glass casing of the lamp rather than filament/refraction due to the glass casing | 1 |
| 1 a ii | Difficulty when the image of the lamp is most clear. | 1 |
| 1 bi | When the image is larger and dimmer the point of greatest sharpness/clarity is harder to judge. | 1 |
| 1 b ii | If the absolute uncertainty increases in proportion to distance the ratio absolute uncertainty/distance will remain constant. | 1 |
| 1 c | $\begin{aligned} & \frac{1}{f}=\frac{1}{v}-\frac{1}{u}=\frac{1}{0.260}-\left(-\frac{1}{1.000}\right)=\frac{63}{13} \mathrm{~m}^{-1} \\ & f=\frac{13}{63}=0.206 \mathrm{~m} \end{aligned}$ | 1 |
| 1 d | Calculating lowest value of $f$ as 0.200 highest value as 0.212 <br> range $=+/-0.006 \mathrm{~m}$ <br> which is greater than 5 mm | 1 1 1 1 |
| 1 e | Equation is of the form $y=m x+c$ <br> $y$ identified with $\frac{1}{v}, x$ identified with $\frac{1}{u}$, and $c\left(y\right.$-intercept) with $\frac{1}{f}$ | 1 |
| 1 f | $\frac{1}{f}=5.0 \mathrm{~m}^{-1} \text { so } f=0.20 \mathrm{D}$ | 1 |
| 1 g | Best-fit line highlights the presence of outliers. <br> A simple mean could be incorrectly weighted by such data points. | 1 |

